

Hydrogen injectivity and recovery in porous media

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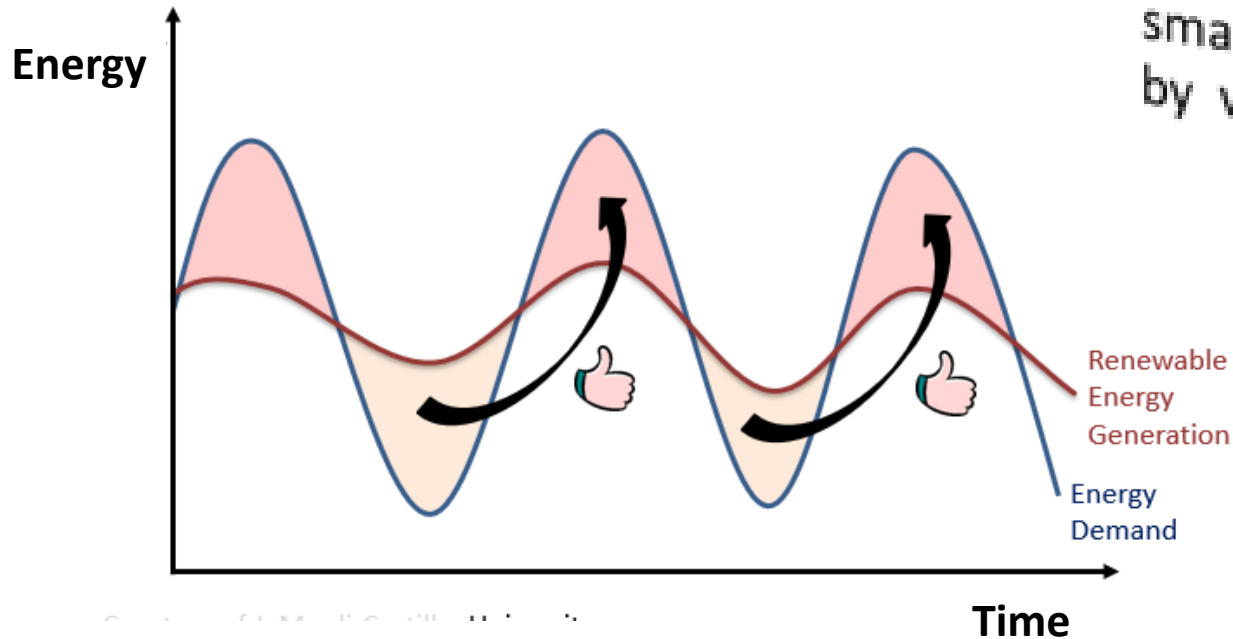


Why porous media storage of hydrogen?

If electricity is fully renewable by 2050, the UK will need 60 TWh storage

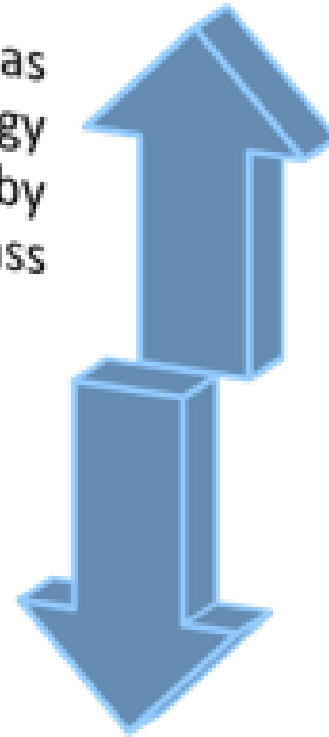
[Howard 2019]

The Renewable Problem



Hydrogen has high energy density by mass

But very small mass by volume



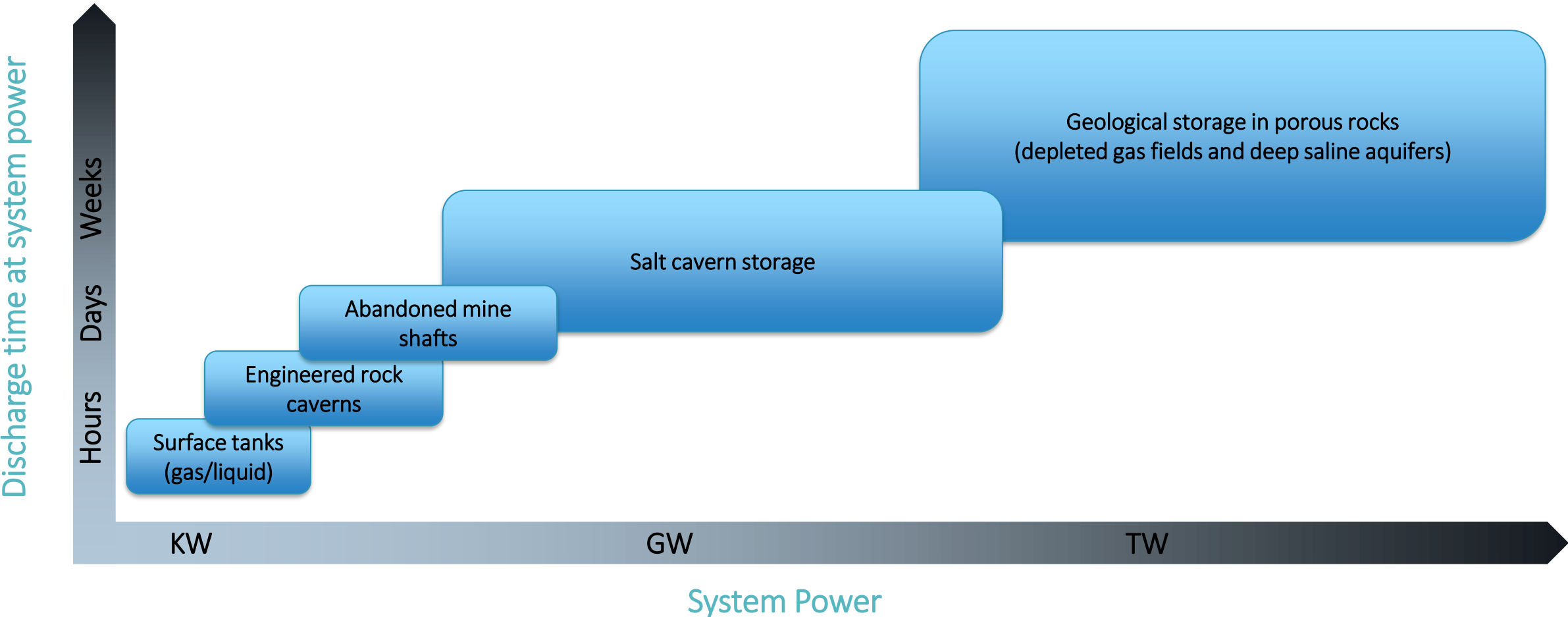
To store the same amount of energy with H₂ we need **3x** more space than for CH₄

Large scale storage is required

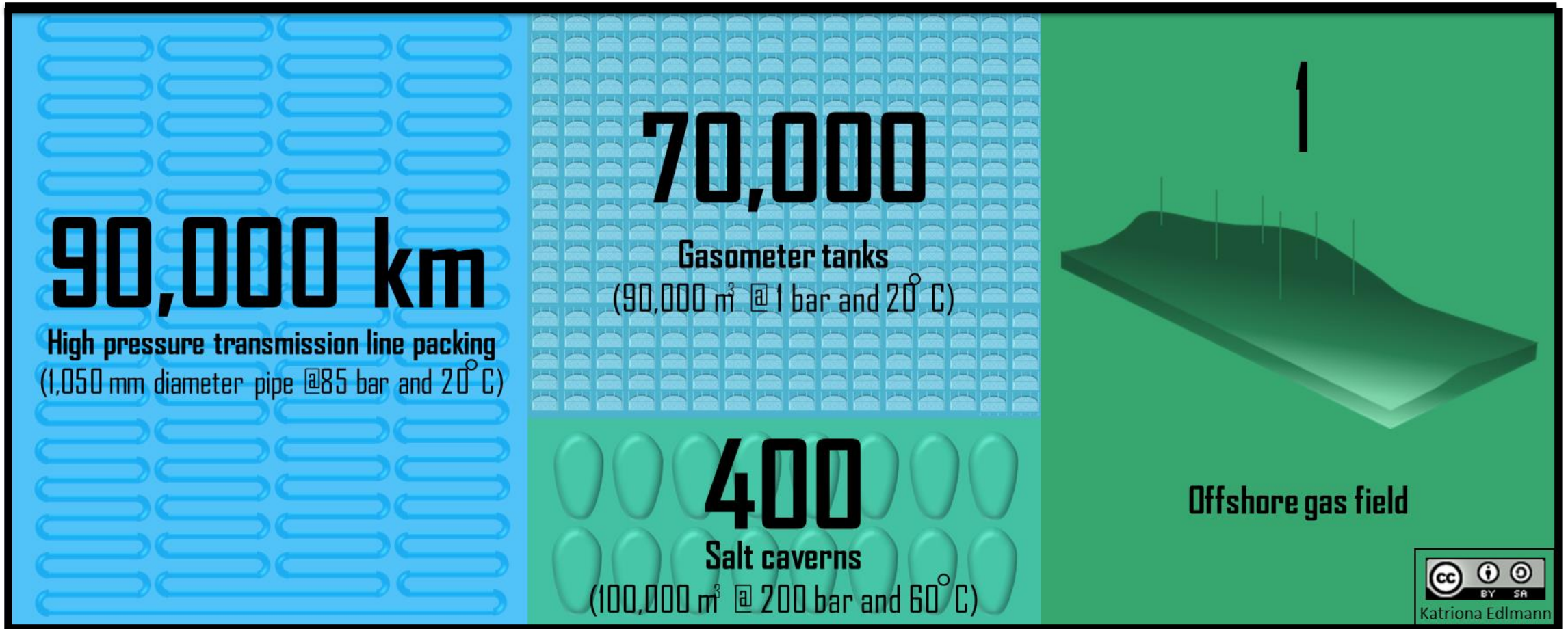
Compressing or liquifying H₂ @ -252.9°C increases the volumetric energy density

Howard, R. (2019). 2050 perspective – Future European power system and strategic implications. Aurora Energy Research

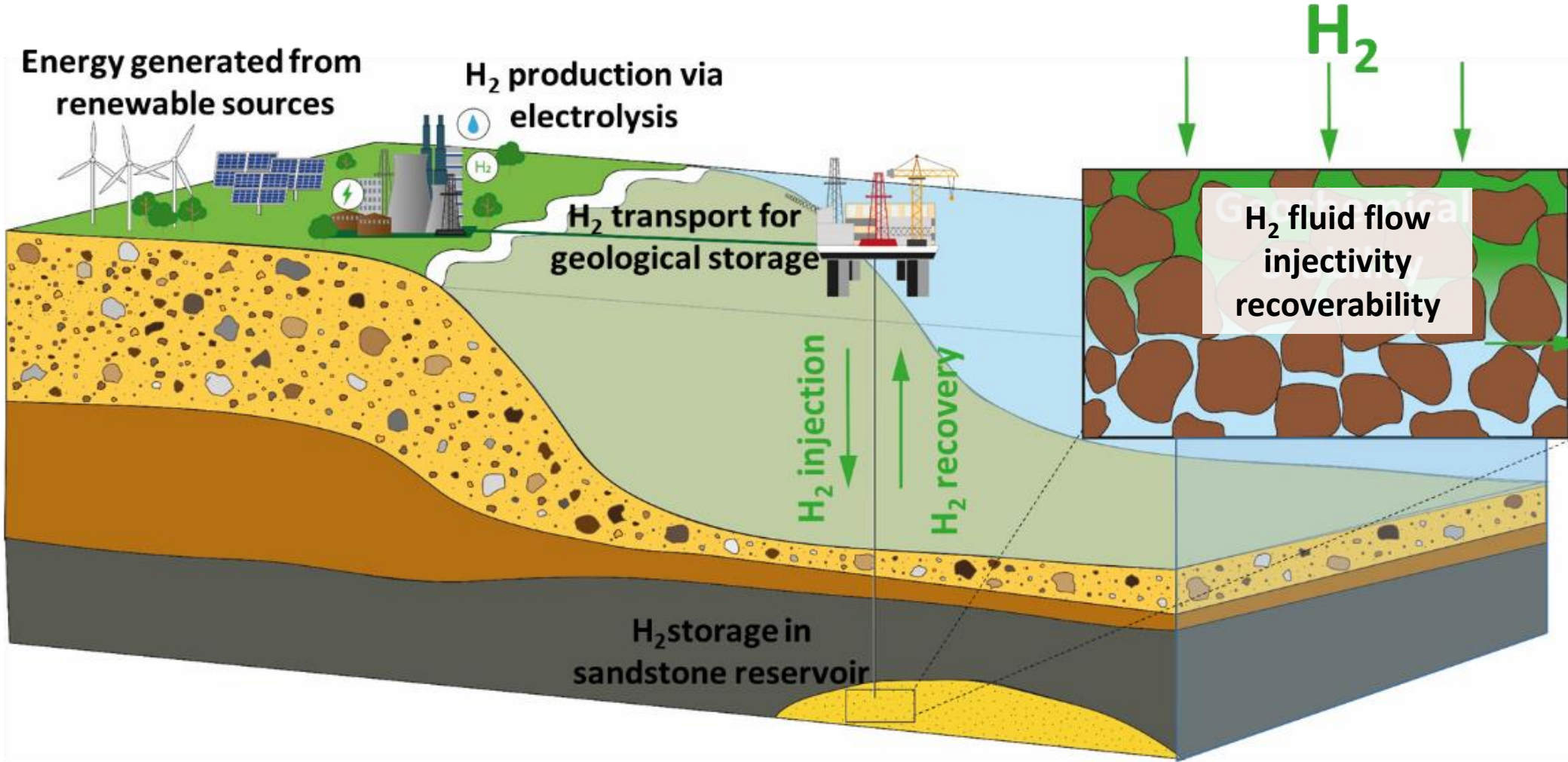
Scales and deliverability of hydrogen storage



Where can we store 20 TWh of H₂?

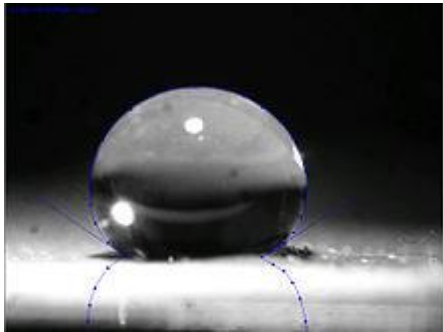


Interseasonal porous media H₂ storage— a technology in its infancy shoes



Parameters that affect H₂ injectivity and recovery

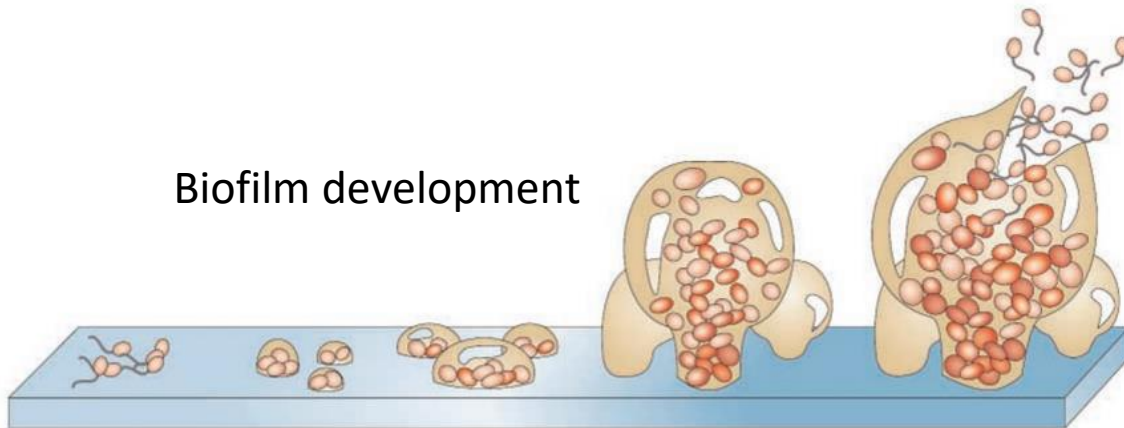
Key parameter: Wettability (contact angle between H₂, brine and rock)



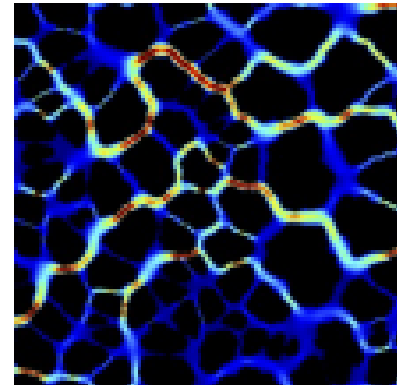
organic acids, oil and biofilm coating affect the wettability and, depending on measurement technique (buoyancy, capillary and gravitational forces acting) salinity, temperature, pressure

- Porosity, permeability, tortuosity, pore network connectivity also important descriptors for H₂ fluid flow *beware microorganisms!*
- Cushion gas

Biofilm development



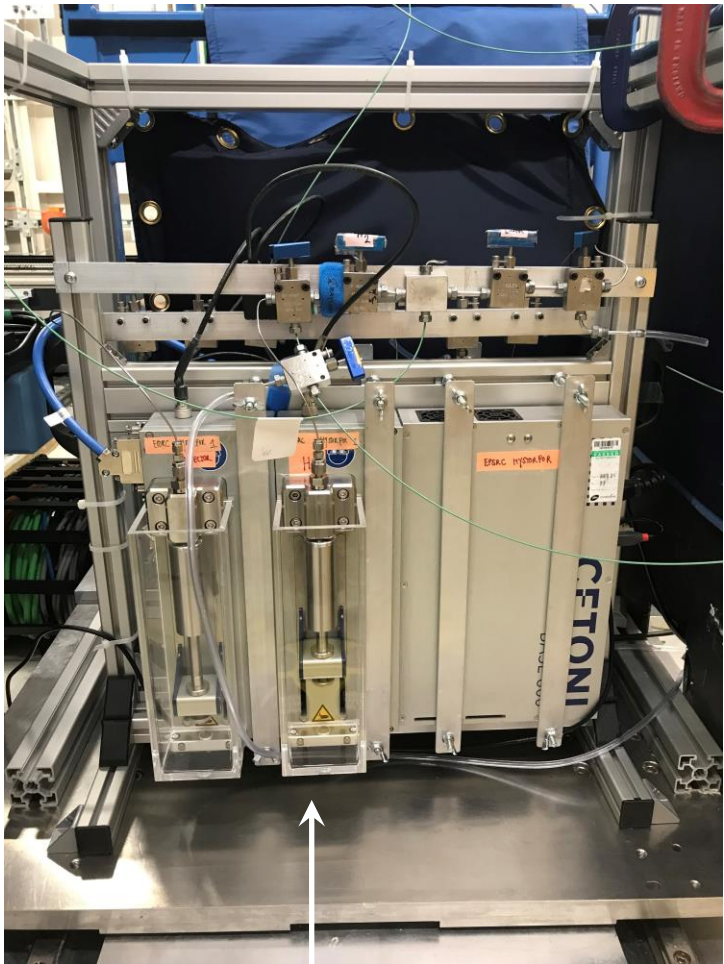
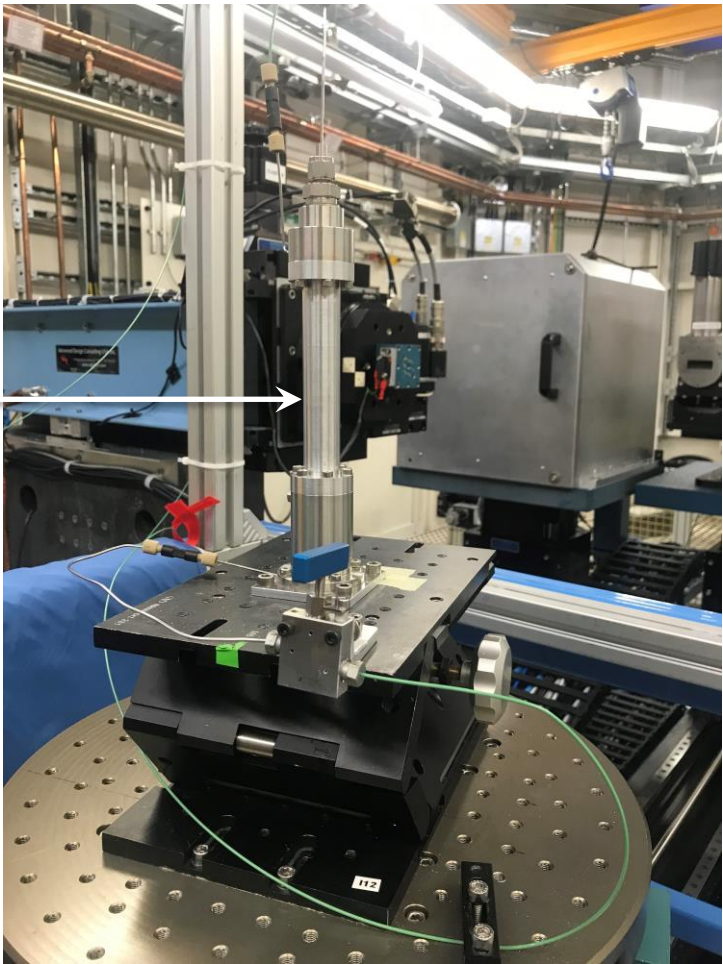
Davies, D. (2003). Understanding biofilm resistance to antibacterial agents. *Nature Reviews Drug Discovery* (2), 114–122



Clogging of pores and pipes

Experiments on H₂ injectivity and recovery at Diamond Light Source, UK

pressure vessel



Rock sample inside X-ray transparent pressure vessel

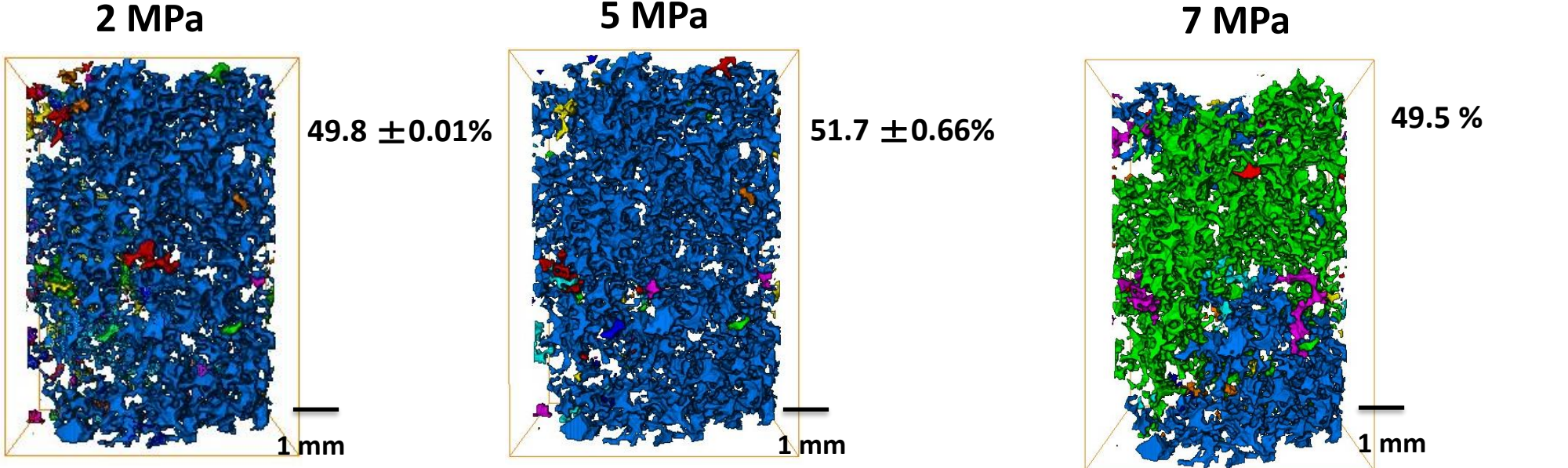
Pumps for injecting H₂ and brine and for the confining pressure and backpressure

Residual H₂ saturation increases with pore fluid pressure

(all at constant confining pressure of 8 MPa)

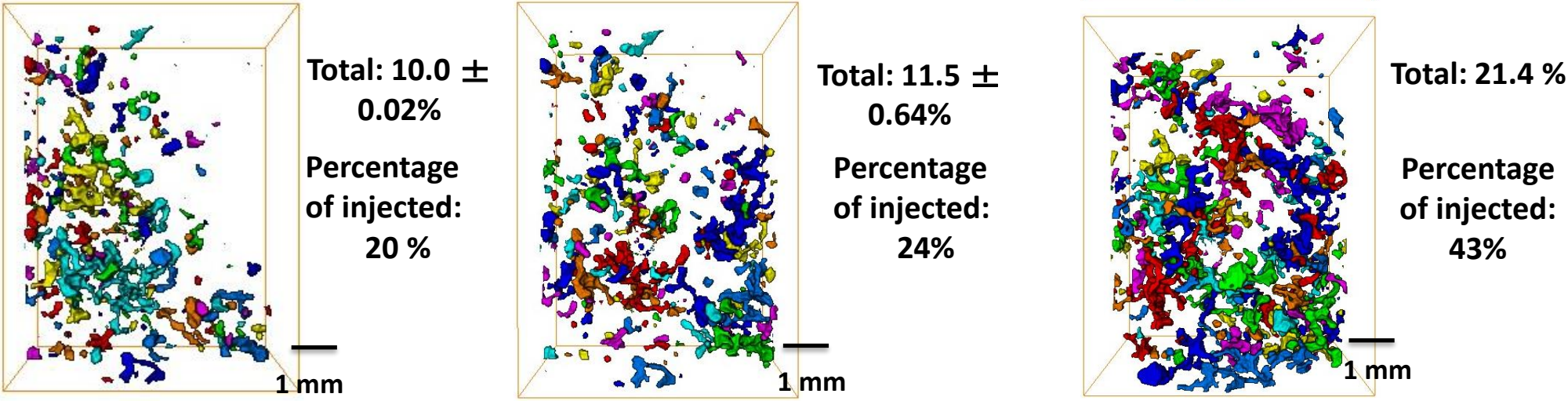
H₂ saturation
after injection
into brine
saturated rock

@C_a = 1.7*10⁻⁸



Residual H₂
saturation
after brine imbibition

@C_a = 2.4*10⁻⁶



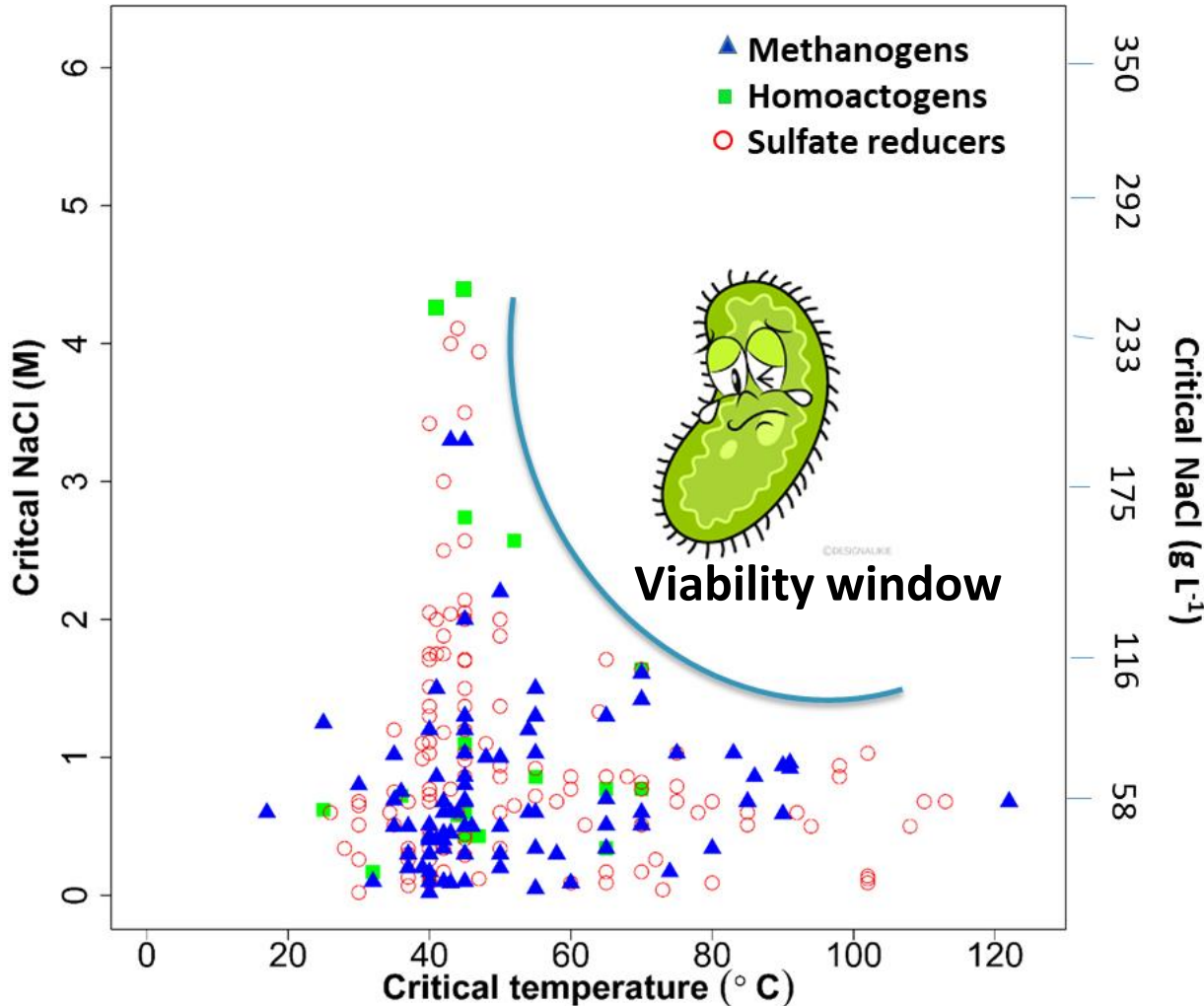
How do these numbers compare to other studies?

Rock dependency
Experimental condition
Methodology

H ₂ Injectivity	Residual H ₂ (recovery)	Rock material	Method	Reference
4 %	Less than 2 % trapped (more than 50 % recovered)	Fontainebleau sandstone	Nuclear Magnetic Resonance at 0.4 MPa and ambient temperature	Al-Yaseri et al. (2022)
65 %	41 % trapped (39 % recovered)	Gosford sandstone (very short sample**)	Micro-CT ambient temperature and pressure	Jha et al. (2021)
36 %	25 % trapped (30 % recovered)	Bentheimer sandstone	Micro-CT 10 MPa and 50°C	Jangda et al. (2022)
50 %	10-21 % trapped (57-80 % recovered)	Clashach sandstone	Micro-CT 2-7 MPa and ambient temperature	Thaysen et al. (2022)

How can we avoid reduced injectivity and withdrawal by clogging ?

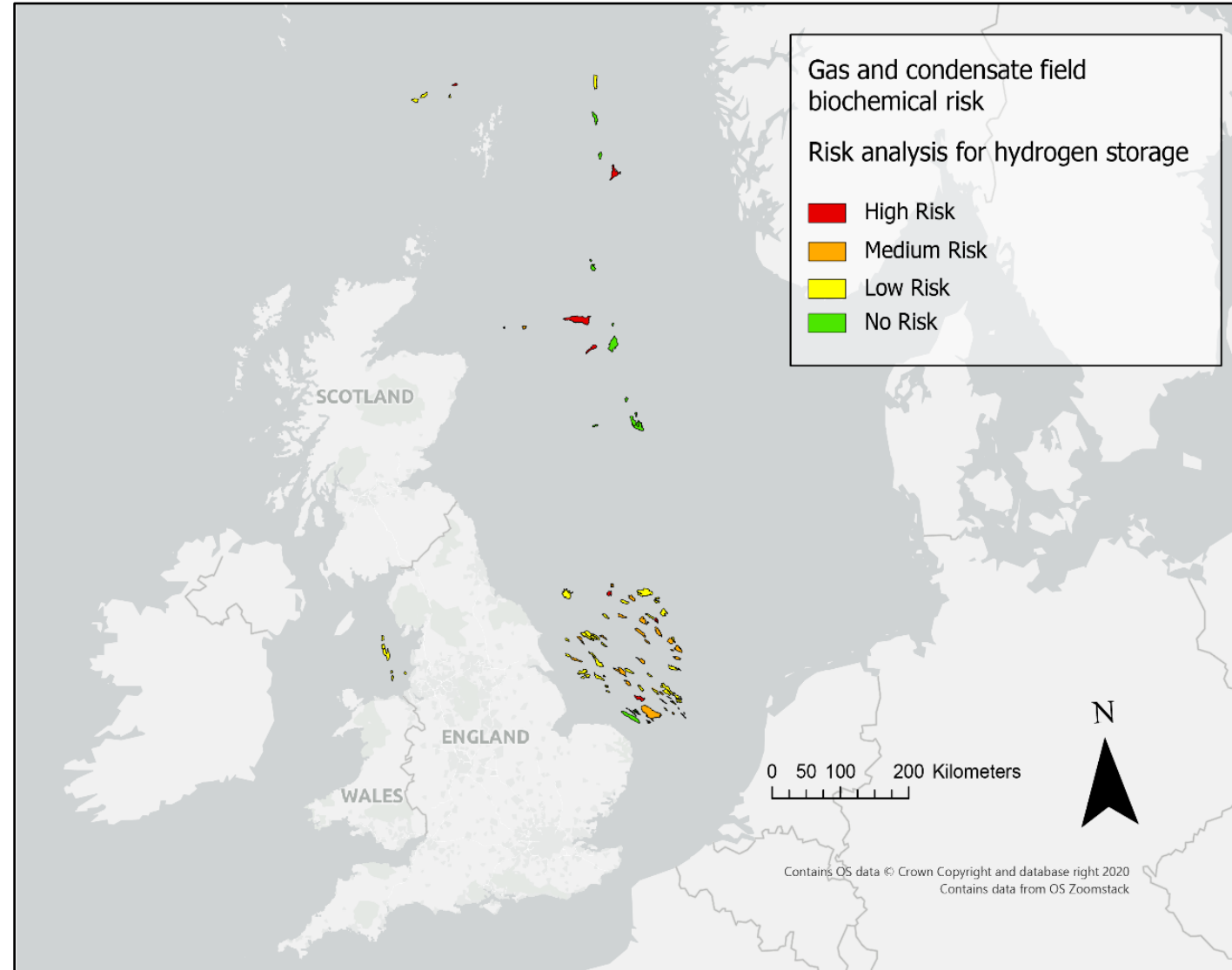
Site selection: Growth criteria of major cultivated H₂ consuming microbes



- Microbial life limits with regards to temperature, pH and salinity for four key hydrogen consuming bacteria:
 - **Methanogens** (consume hydrogen/produce methane)
 - **Homoactogens** (consume hydrogen/produce acetone)
 - **Sulphur species reducing** (consume hydrogen/produce hydrogen sulphide)
- Conditions are unfavourable to bacterial activity:
 - Above temperatures of 122°C
 - Above salinities of 4.4 M NaCl

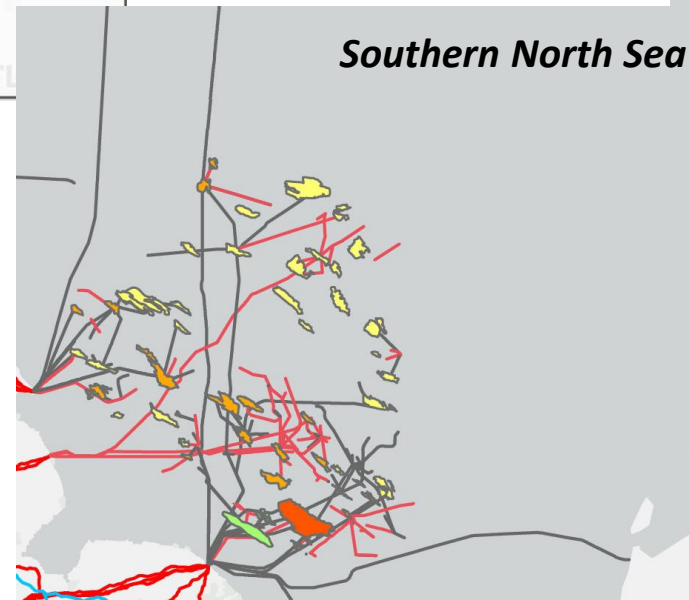
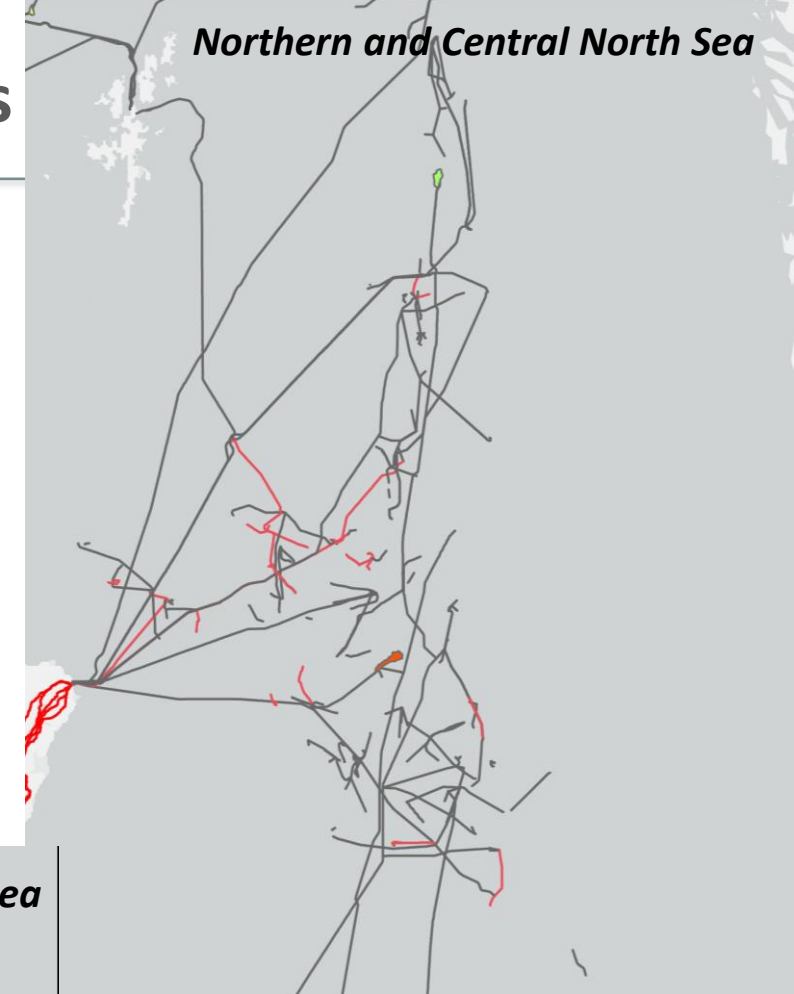
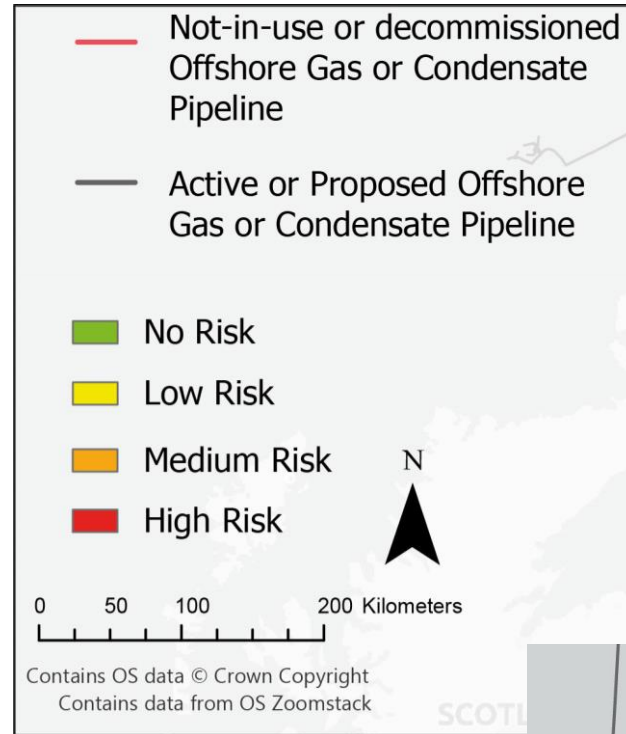
Microbial risk site screening of UKCS depleted gas fields and condensate fields

- **No risk**: fields with a temperature $>122^{\circ}\text{C}$ can be considered as sterile, as no H_2 consuming bacteria have been found above this temperature. **9 UKCS gas fields**
- **Low risk**: fields $>90^{\circ}\text{C}$ are considered paleosterile. **35 UKCS gas fields**
- **Medium risk**: fields $>55^{\circ}\text{C}$ and a salinity $> 1.7 \text{ mol L}^{-1} \text{ NaCl}$, as no cultivated H_2 consuming bacteria can grow in this combination. **22 UKCS gas fields**
- **High risk**: fields $<55^{\circ}\text{C}$ and $< 1.7 \text{ mol L}^{-1} \text{ NaCl}$ because these are conditions optimal for growth. **9 UKCS gas fields**



Microbial risk in depleted gas fields & not-in-use pipelines

Southern North Sea holds many not-in-use pipelines which could be repurposed for H₂ transport to 'no risk' or 'low risk' depleted gas fields



- H₂ injectivity ~4-65% of the pore space and independent of pore fluid pressure
- 30-80 % of the injected H₂ can be recovered making the H₂ storage operation feasible
- H₂ recovery decreases with pore fluid pressure, indicating that shallow reservoirs are more favourable for H₂ storage
- H₂ storage sites should be carefully selected with respect to temperature and salinity as microbial activity can reduce the injectivity and recovery (& consume H₂)

Thank you!

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No show stoppers... so far

- ✓ Perspective paper on enabling large-scale hydrogen storage in porous media – the scientific challenges <https://doi.org/10.1039/D0EE03536J>
- ✓ Biological site screening: We suggest that storage reservoirs over 122°C or with salinities above 4.4 M NaCl equivalent will be less favourable to microbial growth <https://authors.elsevier.com/c/1dYWP4s9Hw2Eu4>
- ✓ No significant geochemical reactions have been observed in our reactive experiments <https://pubs.acs.org/doi/full/10.1021/acsenergylett.2c01024>
- ✓ Column height calculations indicate hydrogen will have a higher column height than methane and that this increases with increasing depth. <https://doi.org/10.1021/acsenergylett.1c00845>
- ✓ Developed an online tool to provide high accuracy thermodynamic property estimations of hydrogen mixtures (CO₂, N₂, CH₄, natural gas) over a range of temperatures and pressures. <https://www.nature.com/articles/s41597-020-0568-6>
- ✓ Cushion gas will play an important role in controlling both injectivity and productivity during hydrogen storage. <https://doi.org/10.1016/j.ijhydene.2021.09.174>
- ✓ Significant storage capacity in depleted gas fields, minimising subsurface competition with other low carbon geoenergy applications such as CCS or CAES. <https://doi.org/10.1016/j.apenergy.2020.116348> and <https://doi.org/10.1021/acsenergylett.1c00845>